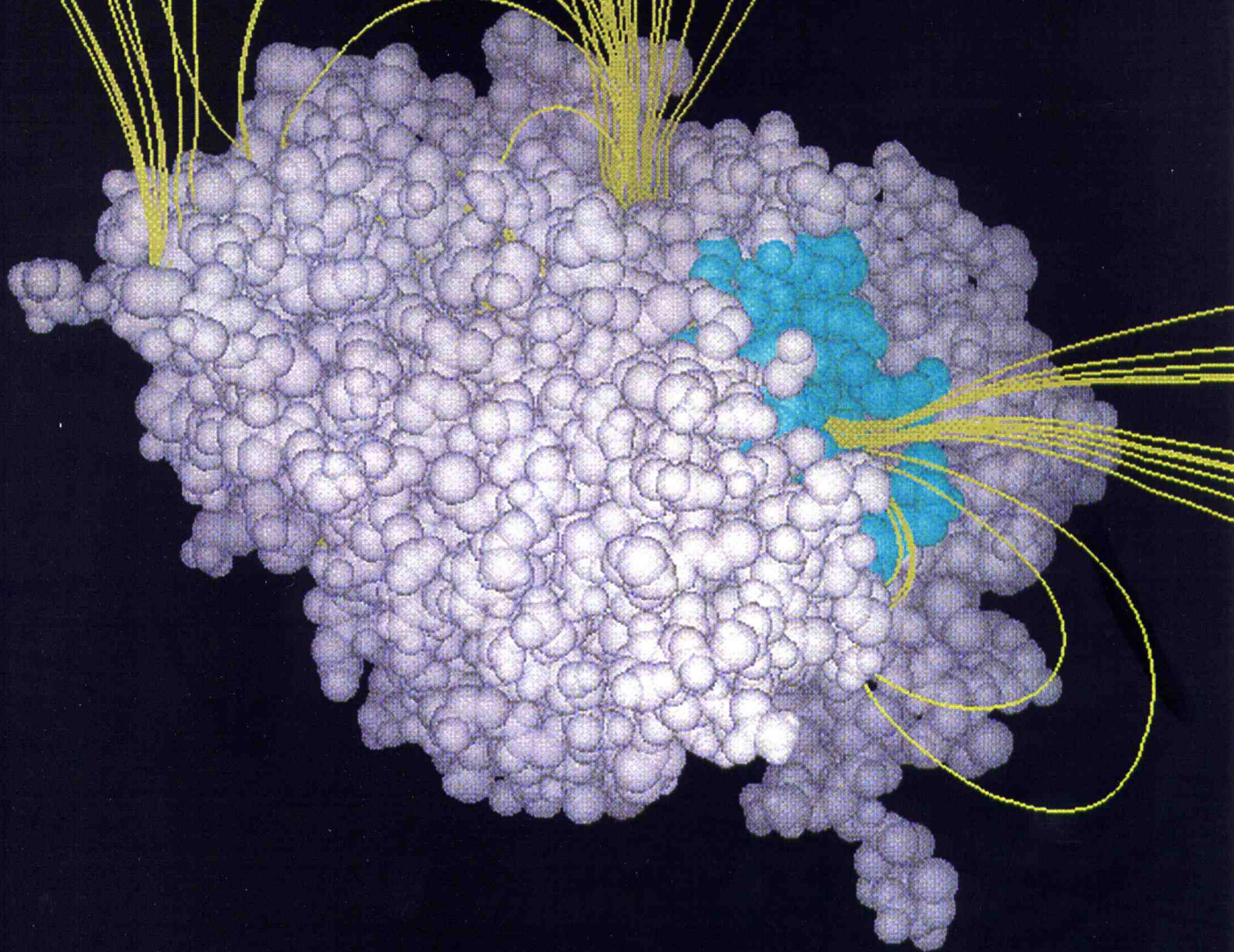


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Model of the enzyme acetylcholinesterase with electrostatic lines of force (yellow) emanating from the interior of the active site (buried). The field lines exit mostly through the main entrance (blue atoms) or through a channel ("back door") to the active site. A

molecular dynamics simulation shows transient opening of this back door (top), which may facilitate the entry or exit of substrate, products, or solvent. See page 1276. [Graphics: Michael K. Gilson, using Insight II software (Biosym)]

Open "Back Door" in a Molecular Dynamics Simulation of Acetylcholinesterase

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The enzyme acetylcholinesterase generates a strong electrostatic field that can attract the cationic substrate acetylcholine to the active site. However, the long and narrow active site gorge seems inconsistent with the enzyme's high catalytic rate. A molecular dynamics simulation of acetylcholinesterase in water reveals the transient opening of a short channel, large enough to pass a water molecule, through a thin wall of the active site near tryptophan-84. This simulation suggests that substrate, products, or solvent could move through this "back door," in addition to the entrance revealed by the crystallographic structure. Electrostatic calculations show a strong field at the back door, oriented to attract the substrate and the reaction product choline and to repel the other reaction product, acetate. Analysis of the open back door conformation suggests a mutation that could seal the back door and thus test the hypothesis that thermal motion of this enzyme may open multiple routes of access to its active site.

The enzyme acetylcholinesterase (AChE) terminates signaling at cholinergic synapses by rapid hydrolysis of the neurotransmitter acetylcholine (ACh). The crystal structure of AChE (1) raises questions regarding substrate entry and product release. First, the active site gorge is very deep and appears too narrow to admit ACh (1, 2). However, the demonstration that quaternary amines enter the active site of crystallized AChE (3) proves that the protein is sufficiently flexible to admit substrate by some route. Second, the inward electrostatic field at the gorge, which is likely to accelerate penetration of the positively charged substrate (4, 5), would seem to impede the exit of the product choline from the mouth of the active site. A thin wall near the base of the

active site, at residues Met⁸³ and Trp⁸⁴, could offer an alternative route for the escape of products (4). This back door might also provide a vent for water molecules during the passage of substrate, products, or both through the narrow gorge. We have studied these issues by performing a molecular dynamics (MD) simulation of AChE and analyzing the resulting time series of protein conformations with regard to active site accessibility.

Unobserved atoms were added to the *Torpedo californica* AChE dimer (1), as described in (5), using the program Quanta (Molecular Simulations, Waltham, Massachusetts). Protonation states at neutral pH were assigned on the basis of atomic solvent accessibilities, of salt-bridging and hydrogen-bonding opportunities, and of the putative catalytic mechanism, which requires the active site His⁴⁴⁰ to be neutral (6). The model enzyme was immersed in water, and the classical laws of motion solved numerically with weak coupling to a heat bath at 300 K to generate 119 ps of analyzable conformations (7).

We searched for conformations in which the active site had any aperture large enough to pass a water molecule, generating for each conformation a Lee and Richards (8) solvent-accessible dot surface with a 1.4 Å probe. The molecular

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surface was separated into the outer surface and some number of cavity surfaces, based on distances between surface points (9). Of these surface subsets, the active site surface was defined as that contacted by Glu¹⁹⁹ OE2, because this atom was consistently at the surface of the active site interior. Conformations with an open active site were those in which the numbers of connected points in both the active site and outer surfaces were identical.

Conformations with an open back door were sought as follows. For each conformation having an open active site, the active site entry was blocked by the deletion of all surface points within 8 Å of a point located approximately in the middle of residues Glu⁷³, Asn²⁸⁰, Asp²⁸⁵, and Leu³³³ that defines the entrance to the gorge. If the active site surface was still continuous with the outer surface of the enzyme, an alternative aperture existed.

This surface analysis detected a transient opening, occurring in the first 20 ps of production MD and persisting for 0.3 ps. A channel (Fig. 1) formed in the thin active site wall at Trp⁸⁴, the side chain of which probably contributes to binding of the quaternary ammonium of ACh (1, 3, 10). The channel began near Trp⁸⁴ CH2 and CZ2, Gly⁴⁴¹, and the ring of Tyr⁴⁴² and curved around the edge of Trp⁸⁴ to emerge at a surface dimple near Glu⁴⁴⁵. Its path led directly through a solvent-sized cavity that is observed in the crystal structure and which lies near Val¹²⁹. The displacement of the Trp⁸⁴ indole, relative to its location in the crystal structure, resembles that of a camera shutter because it is completely in-plane.

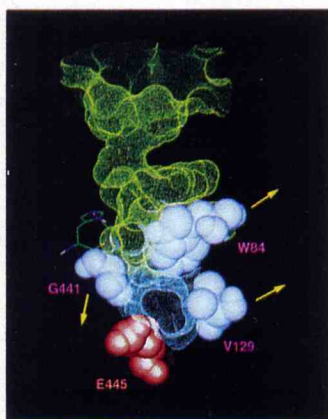


Fig. 1. Active site gorge with open back door. Green dots, solvent-accessible surface of active-site gorge (T, top of gorge); white dots, channel to back door. Space-filling atoms are shown for Trp⁸⁴ (W84), Val¹²⁹ (V129), Gly⁴⁴¹ (G441), and Glu⁴⁴⁵ (E445). Vectors indicate directions of displacement of three of these residues relative to their positions in the crystal structure. Stick diagrams indicate catalytic residues His⁴⁴⁰ and Ser²⁰⁰.

The channel can be opened by shifting only residues Trp⁸⁴, Val¹²⁹, and Gly⁴⁴¹ from their crystal coordinates to their positions in the open conformation. The average atomic displacement for these residues, relative to the crystal conformation, is only 1.3 Å. The Trp⁸⁴ residue is somewhat more mobile than average: The average root-mean-square (rms) fluctuation of its nonhydrogen atoms is 1.2 Å (main chain 0.8 Å, side chain 1.4 Å), compared with 0.72 Å for all nonhydrogen atoms within 25 Å of the dynamical center. The Val¹²⁹ and Gly⁴⁴¹ residues have unremarkable rms fluctuations of 0.70 Å.

Several points are significant to this opening mode. First, the time required for AChE to hydrolyze one substrate molecule is about 0.1 ms (11). Thus, although the observed opening event is brief, that it occurs at all in a 119-ps simulation suggests that the back door can play a functional role on the longer time scale of catalysis. Second, the small size of the atomic displacements required to open the back door means that the open conformation should be easy to achieve; experiments have demonstrated that protein structures fluctuate substantially (12). Third, the presence in

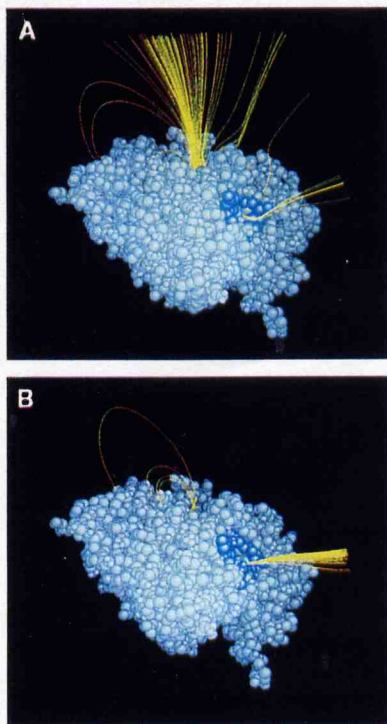


Fig. 2. Electric field lines for AChE with (A) open and (B) closed back door. The field lines (yellow) are traced from the region of negative potential near Glu¹⁹⁹ OE2 deep in the active site. Blue atoms highlight the opening to the active site gorge. The back door is clearly marked by the vertical bundle of field lines in (A). Field lines were computed and displayed with GRASP (16). The orientation is changed relative to Fig. 1.

the crystal structure of a solvent-sized cavity in the thin wall of the active site where the channel forms suggests that the x-ray data may reflect similar opening events in a time-averaged fashion. Finally, no opening events were observed in other portions of the active site wall, although every stored conformation was analyzed. Thus, Trp⁸⁴ defines the only permeable part of the active site wall in this simulation.

Because AChE appears to generate a functionally important electrostatic field (4, 5), we examined the effect of the open back door on this field (13). When the back door is open, field lines initiated at the active site Glu¹⁹⁹, near Trp⁸⁴, exit preferentially through this aperture (Fig. 2A). However, for the crystal structure most of these field lines emerge from the mouth of the active site gorge (Fig. 2B). When the electrostatic potential was evaluated at each point on a Lee and Richards-type surface (8), the surface point with the lowest electrostatic potential lay at the bottom of the active site, near Ile⁴⁴⁴. Therefore, ACh and choline are attracted to the base of the gorge, either through the front or the back door. This observation militates against the idea that the same field that guides ACh into and down the gorge urges choline out the back door. However, the field accelerates the departure of the other product, acetate ion. Also, a more widely opened back door might significantly weaken the active site fields by increasing exposure of the active site to the high dielectric solvent. This effect would facilitate the release of choline.

One way to test whether a functionally important back door does exist would be a kinetic study of a mutant enzyme whose back door is sealed. Of the three residues whose movement opens the channel, Trp⁸⁴, Val¹²⁹, and Gly⁴⁴¹, it appears that only Val¹²⁹ could be modified without the disruption of enzyme function by undesired mechanisms. Steric analysis of the crystal structure suggests that a Lys or Arg would fit well at position 129, with the positive group between Met⁸³, Asp¹²⁸, Glu⁴⁴⁵, and Leu⁴⁵⁶ (Fig. 1). In this position, the side chain would fill the surface dimple where the back door channel emerges. The positive group of the mutant side chain would be far (~11 Å) from the interior of the active site. To our knowledge, no special role has been suggested for Val¹²⁹, and because it lies at the protein surface, it appears to be structurally unimportant. However, aligned cholinesterase amino acid sequences show little variation at this position: Eight have Val and two have Ile (14). Isoleucine seems unlikely to block the back door because it possesses only one more methylene than Val. If the mutation of Val¹²⁹ does slow the enzyme, it will be necessary to confirm the absence of structural distortion through crystallographic studies.

Experiments have shown that thermal fluctuations make proteins somewhat porous, especially to small, nonpolar molecules (15). Our analysis supports the concept that the active site of AChE has a particularly porous wall at Trp⁸⁴, which may be of functional importance. Kinetic energy gained by the catalytic residue His⁴⁴⁰ during hydrolysis may pass by way of the peptide linkage to Gly⁴⁴¹, one of the channel residues. This energy might increase the probability of opening, causing an organized sequence of catalysis and channel opening.

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- All Arg, Lys, Asp, Glu, NH₂-terminal, and COOH-terminal groups were assigned "ionized" (except Glu⁴⁴³). The Glu⁴⁴³ residue was assigned "neutral" (OE2 protonated) because it is fully desolvated and appears to form only two to three hydrogen bonds. All Tyr and Cys side chains were set neutral. Histidine protonation sites were His³, ND1; His²⁶, ND1; His¹⁵⁹, NE2; His¹⁸¹, ND1, NE2; His²⁰⁹, NE2; His²⁶⁴, ND1; His³⁶², ND1; His³⁹⁸, ND1; His⁴⁰⁶, ND1, NE2; His⁴²⁵, ND1, NE2; His⁴⁴⁰, ND1; His⁴⁷¹, ND1, NE2; His⁴⁸⁶, NE2; and His⁵¹³, ND1, NE2. The net charge per monomer was -5. Both monomers were assigned identical protonation states. Polar hydrogens were added with the ARGOS package (17).
- With only crystallographic solvent included, the energy was minimized with respect to hydrogen positions for 200 steepest descent steps. The energy was then minimized for 200 steepest descent steps with respect to the coordinates of residues 535 to 537 and 1072 to 1074 to form the COOH-terminal disulfide bond. A pre-equilibrated box of waters [SPC/E model (18)] was overlaid repeatedly on the system to fill in a 40 Å sphere centered on atom NE2 of the catalytic His⁴⁴⁰. The 142 crystallographic waters were retained. Waters closer than 2.5 Å to any crystallographic heavy atom were rejected. Atoms more than 35 Å from His⁴⁴⁰ NE2 were fixed in space for all subsequent calculations to create a constant-volume dynamical system containing 5252 protein atoms and 3117 water molecules. Energy was minimized with respect to water coordinates for 200 steepest descent steps. Then MD on water only was performed for 20 ps, with velocity reassignment at 300 K every 0.2 ps and velocity rescaling (19) with time constant 0.2 ps. Energy was then minimized with respect to protein coordinates for 200 steepest descent steps. MD on the protein only was performed for three segments of 5 ps each, with velocity reassignment every 0.2 ps and rescaling with time constant 0.1 ps, at 100 K, 200 K, and 300 K, respectively. Before production calculations were made, MD was used to equilibrate the entire system for 20 ps at 300 K, with separate solute and solvent velocity rescaling with time constant 0.2 ps. During the first 20 ps of production, coordinates were stored every 20 steps. Subsequently, coordinates were stored every five steps. All interactions in a short-range (10 Å) pair list were updated each step, while all interactions in a long-range (12 Å) pair list were updated every five steps. Separate solvent-solvent, solvent-solute, and solute-solute pair lists were updated every 10, 15, and 20 steps, respectively. Pair lists were based on charge groups. SHAKE (20, 21) was used to constrain bond lengths, permitting the use of a 2-fs time step. All calculations were made with the ARGOS program package (17), with GROMOS atomic parameters (22). System temperatures approximated 301 K, with solute temperatures of about 298 K and solvent temperatures of about 302 K. Total system potential energy averaged -200×10^3 kJ mol⁻¹, with a drift of -8.0 kJ mol⁻¹ ps⁻¹ during the final 100 ps. During the final 100 ps, the rms atomic position deviation from crystal coordinates of nonhydrogen atoms in the innermost 25 Å of the simulation sphere averaged 1.7 Å, with a residual upward slope of 0.001 Å ps⁻¹.
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