# Crystal Structure of Demetallized Concanavalin A: the Metal-binding Region 

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#### Abstract

We have determined the crystal structure of demetallized concanavalin A, at a resolution of $3 \cdot 2 \AA$, by molecular replacement using the known structure of native concanavalin A. Refinement of the initial model using a constraint-restraint re-ciprocal-space least-squares procedure caused the conventional crystallographic agreement ( $R$ ) factor to decrease from 0.47 to a final value of $0 \cdot 26$. There are significant conformational changes in the metal-binding region involving residues Asp19 and His24, which are substantially closer to each other than in native concanavalin A. These residues form an internal salt bridge which does not exist when the metal ions are attached to the protein. The binding site for transitionmetalions is still intact, but the calcium site is not, since one of its two carboxylic ligands, Asp 19, is unavailable. Flexibility is observed for one of the transitionmetal ligands, Glu8, as well as for some segments of the backbone. The latter could account for the increased susceptibility of demetallized concanavalin A to proteolysis.


## 1. Introduction

Concanavalin $A$ is a saccharide-binding protein of the Jack bean (Sumner, 1919). Its saccharide-binding properties in solution have been shown to depend on the presence of two metal ions in the protein (Sumner \& Howell, 1936; Yariv et al., 1968). One of these is bound at a site (Sl) with specificity for divalent metal ions having affinity for nitrogen ligands (e.g. $\mathrm{Mn}^{2+}, \mathrm{Co}^{2+}, \mathrm{Zn}^{2+}$ ). Occupation of Sl by a metal ion is attended by the formation of a second metal-binding site (S2) with high specificity for $\mathrm{Ca}^{2+}$ (Kalb \& Levitzki, 1968; Shoham et al., 1973).

Spectroscopie studics of concanavalin A-metal complexes indicate that binding of the $\mathrm{Ca}^{2+}$ ion at S 2 leads to a rather rigid organization of the transition metal-binding site, Sl (Nicolau et al., 1969; Meirovitch \& Kalb, 1973; Barber \& Carver, 1973; Kalb \& Pecht, 1973; Meirovitch et al., 1974; Goldammer \& Zorn, 1974). Enzymic digestibility is found to decrease greatly on occupation of SL; the native form, with both S1 and S2 occupied, is virtually impregnable to proteolytic enzymes (Blumberg \& Tal, 1976).

The details of the structural changes in the protein on successive occupation of the two motal-binding sitcs can be elucidated by comparisons among X-ray structures of metal-free concanavalin A, its transition metal complex, and the native protein containing both metals. Of these, the native form has been crystallized and its X-ray structure has been described in some detail (Edelman et at., 1972; Hardman \& Ainsworth, 1972; Becker et al., 1975).

In the present study, we have determined the structure of demetallized concanavalin A. The crystals, which have been described elsewhere (Jack et al., 1971), have the space group $P 2_{1} 22_{1}$ with axial lengths $a=84 \cdot 3 \AA, b=91 \cdot 2 \AA, c=61 \cdot 1 \AA$ and are related to crystals of the native form, with space group I222, by a small rotation of the structure about the crystallographic $y$-axis. These crystals are well suited to such a comparative study because they can be converted, in the solid state, to crystals of the native form (1222) simply by soaking them for several days in a dilute solution of $\mathrm{Mn}^{2+}$ and $\mathrm{Ca}^{2+}$.

The unit-cell dimensions and protein content are nearly identical in the two crystal forms. However, in the native crystal the asymmetric unit is a single polypeptide subunit of $M_{\mathbf{r}} 26,000$ whereas in the demetallized crystal it is a dimer of $M_{r}$ 52,000 . The similarity between these crystals prompted us to use the molecularreplacement method (Rossmann, 1972) for the structure determination.

X-ray studies of native concanavalin A have identified the amino acid side-chains which participate in the two metal-binding sites. The amino acids involved are included in the sequence of residues 7 to 35 from the $\mathbf{N}$-terminus. In this study, we have focused our attention on this region, shown schematically in Figure 1. The differences which we find between the native and demetallized proteins form the structural basis for a mechanistic model of the interactions in solution between the two metal-binding sites, which we have reported briefly elsewhere (Shoham et al., 1978).

## 2. Experimental

(a) Data collection and processing

Demetallized concanavalin A was prepared and crystallized as previously deseribed (Kalb \& Levitzki, 1968; Jack et al., 1971). These crystals diffract out to a minimum spacing of about $3 \AA$.


Fig. 1. Schematic view of the $\mathrm{Mn}^{2+}$ and $\mathrm{Ca}^{2+}$ binding region in native concanavalin $A$ (based on co-ordinates kindly supplied by Dr K. Hardman). The open lines represent virtual bonds betwoon succossive alpha-carbon atoms of residues 7 to 35 . The side-chain ligands are shown in solid lines, and the co-ordination of the metal ions in dashed lines. Water molecules are designated "W". Note that all the direct protein ligands are within the loop of residues 8 to 24 .

Intensity measurements were made on a Nonius CAD-3 automatic diffractometer, fitted with a $40-\mathrm{cm}$ helium-filled tube between crystal and counter, and using Ni -fillered $\mathrm{CuK} \alpha$ radiation. Measurements were made with an $\omega$-scan technique (Watson et al., 1970). The scan width was chosen wide enough to cover at least 1.5 times a complete profile of each reflection, and the positions of the reflections were determined by maximizing tho integrated intensities over 0.66 of the scan width. Twice the integrated intensity over the remaining points was subtracted as background.

An empirical absorption correction was applied as deseribed by North et al. (1968). Absorption measurements were made several times during the course of data collection and, as no significant changes were observed in absorption curves for the same crystal, these were averaged. Radiation damage and crystal misalignment were monitored by remeasuring 8 standard reflections after every 100 intensity measurements. An empirical correction for radiation damage, based on the decay in the monitor reflections, was applied up to a maximum intensity loss of $\mathbf{1 5 \%}$, after which the crystal was discarded.
Data were collected from 3 crystals in overlapping radial shells. No significant intensities were observed beyond $3 \cdot 2 \AA$. Data from the different crystals were scaled and merged using common reflections (Rollett \& Sparks, 1960). The average deviation from the mean ( $\Sigma||F|-|\bar{F}|| / \Sigma|\vec{F}|)$ was 0.052 for about 1000 structure amplitudes common to crystals 1 and 2 and 0.076 for about 500 common to crystals 2 and 3.

A chloroplatinate derivative of demetallized concanavalin A was prepared by soaking crystals of the protein in a crystallizing medium containing $0.1 \mathrm{~mm}-\mathrm{K}_{2} \mathrm{PtCl}_{4}$ for 7 days. All 3 centric projections of parent and derivative crystals were recorded on $14^{\circ}$ precession photographs at 100 mm specimen-to-film distance with Ni-filtered CuK $\alpha$ radiation, and intensities measured on an Optronics P-1000 microdensitometer. Structure amplitudes were derived with a program kindly supplied by Dr Gerson Cohen.

## (b) Orientation and translation search

The change in the crystallographic space group from I222 in native concanavalin A to $P 2_{1} 22_{1}$ in the demetallized form can be accounted for by a rotation of a rigid tetrameric cluster of protomer subunits about the 2 -fold axis along $y$ (Jack et al., 1971). Thus, the dyads along the $x$ - and $z$-axes in the native form become non-crystallographic in demetallized concanavalin $A$, which has an asymmetric unit composed of two protomers related by a local dyad. Translation of the intact tetramer along the $y$-axis is also consistent with the change in space group, but translation along or rotation about the $x$ - and $z$-axes would imply relative movements of dimers within the tetramer.

We have investigated all these possible motions using, as a search model, a rigid dimer of subunits I and II which, because of their many intermolecular contacts, are believed to constitute the stable dimer in solution (Reeke et al., 1975). Atomic co-ordinates for the search model were kindly supplied by Dr Karl Hardman (Hardman \& Ainsworth, 1975); the metal ions were not included. The orientation and translation search was based on computation of conventional crystallographic agreement $(R)$ factors between observed structure amplitudes and those calculated for the search model in the various possible posilions. Investigations were first made of movements of the intact tetramer by rotation about the $y$-axis or translation along it. Subsequently, after the best position was found for the tetramer ( $7^{\circ}$ rotation about the $y$-axis, see Results), further possible movements of the dimer were investigated in torms of translations along the other 2 axes or combinations of rotations about them.

## (c) Map interpretation and structure refinement

A portion of an initial electron density map is shown in Fig. 2. The coefficients here are $\left(2\left|F_{\mathrm{o}}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right)$, where $\left|F_{\mathrm{c}}\right|$ and $\alpha_{\mathrm{c}}$ are the amplitude and phase of the native stracture, rotated $7^{\circ}$ in this cell, and $\left|F_{0}\right|$ are the observed structure factor amplitudes for tho demetallized structure. The unweighted $R$ factor between these $\left|F_{o}\right|$ and $\left|F_{\mathrm{c}}\right|$ is 0.47 for data to $3 \cdot 2 \AA$ resolution. As might be expected for such a starting model, inspection of the map shows that following chain continuity is difficult and detailed interpretation would be unreliable. To overcome this difficulty we have refined the structure away from the starting model and experimented with the usefulness of different types of maps.


FIa. 2. Two successive 5 A.thick postions of the deetnom density map of dompetalliza concanavalin $A$ in the metal-binding region prior to refinement. The coefficients of the Fourier summation are $\left(2\left|F_{0}\right|-\left|F_{c}\right|\right) \exp \left(i \alpha_{c}\right)$. Contours are drawn from a level corresponding to the standard deviation ( $\sigma$ ) of the electron density map in equal intervals of half that value.

$$
n=\left(\sum_{i=1}^{n_{i}} \rho_{i}^{2} / n\right)^{1 / 2}
$$

where $p_{i}$ is the electron density at each of the $n$ grid points in the non-protein portion of the map, (more than $1 \cdot 5$ A from any protein atom). Superimposed is the initial atomic model. Note tho lack of chain continuity at residues 23 to 25 and the high level of unmatehed density in some parts of the non-protein region.

A constrained-restrained reciprocal-space least-squares refinement method (CORELS) was used. This method was developed for the refinement of the transfer RNA structure at a similar resolution (Sussman et al., 1977), and has now been adapted for protein work (O. Herzberg and J. L. Sussman, manuscript in preparation). In this procedure each amino acid residue is considered to be an independent unit of the structure, but is restrained to have a suitable spatial relationship to neighbouring residues in the chain. Within each unit, dihedral and main-chain bond angles, as well as the composite temperature factors of groups of atoms, may be refined. All restraints are implemented by means of flexible "springs" between pairs of atoms, and by suitably varying the "stiffness" of these springs it is possible to make the distances between pairs of atoms as close to standard lengths as is desired.
The strategy we used in refining the structure (that is, minimizing the differences between the observed and calculated structure factors) was initially to vary only a minimal number of parameters and gradually to allow more to vary in the following order.
(1) Rigid-body movements (rotation and translation) of individual amino acids, with restraints between adjacent residues in the sequence to maintain proper inter-residue geometry.
(2) Variation of the side-chain dihedral angles.
(3) Introduction of "repulsive" springs between non-bonded atoms to minimize total repulsive interaction duc to bad van der Waals' contacts.
(4) Refinement of 2 group temperature factors per residue, 1 for main-chain atoms ( N , $\left.\mathrm{C}_{\alpha}, \mathrm{C}, \mathrm{C}^{\prime}, \mathrm{O}\right)$ and one for the remaining atoms.
This sequence was designed to allow the biggest movements of large groups of atoms to occur first. The asymmetric unit refined consists of the 3612 non-hydrogen and non-metal atoms of the dimer, and the behaviour of equivalent residues in the 2 monomers provided a check of the accuracy of the results.

The usefulness of 3 different maps was investigated. These were:
(1) "Chopped" difference maps, with coefficients ( $\left.\left|F_{\mathrm{o}}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right)$. The calculated structure factor amplitudes and phases excluded contributions from the part of the structure under investigation. These maps have the advantage that no model information is included for the part of the structure under study and so an unbiased interpretation may be made.
(2) A difference map, with coefficients $\left(\left|F_{0}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right)$. This map will have no features where the model is correct, negative features where atoms have been incorrectly placed, and positive features where they have been omitted from the model. Although difficult to interpret in detail at this resolution, this type of map has the advantage of highlighting incorrectly placed features or disorder. Fig. 3 shows partof such a map where a group of atoms have been misplaced in the demetallized structure. Comparison of this with the relatively clean final difference maps demonstrates the usefulness of the map in this regard. Analysis of the average density around individual atomic positions in such maps allows improvement or deterioration in the model during the refinement process to be followed semiquantitatively.
(3) A map with coefficients ( $\left.2\left|F_{\mathrm{o}}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{c}\right)$, which may be written as $F_{\mathrm{c}} \exp \left(i \alpha_{\mathrm{c}}\right)$ $+2\left(\left|F_{0}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right)$, where the 1st term represents an exact calculated map of the current model and the 2nd a map of the difference density between this model and the correct structure; the factor 2 in this term partly compensates for the approximation involved (Luzzati, 1953). Thus the sum of these 2 terms gives approximately the electron density of the correct structure.

## 3. Results

## (a) Orientation and translation search

Figure 4 shows the behaviour of the standard $R$ factor as a function of the rotation of the dimeric asymmetric unit about the $y$-axis. Based on these results, the rotation


Fig. 3. A $5 \AA$-thick portion of a difference electron density map of a part of protomer I, based on a model in which residues 18 to 22 are misplaced. The coefficients of the Fourier summation ar-$\left(\left|F_{0}\right|-\left|F_{c}\right|\right) \exp \left(i \alpha_{c}\right)$. Positive and negative contours are shown in solid and dashed lines, respectively, at oqual intervals of twice the standard deviation (o) of the non-protein region of the map. This map shows how easy it is to deteet inenrrent features in the model, and may bo compared with the much cleaner difference map for the final refined model, shown in Fig. 7.
angle was taken to be $7 \cdot 0^{\circ} \pm 0 \cdot 1^{\circ}$. A systematic investigation of all possible combinations of rotations about the $x$ - and $z$-axes, in $5^{\circ}$ intervals (after $7^{\circ}$ rotation about the $y$-axis), showed only $R$ factors greater than $50 \%$ for 83 strong reflections (cf. Fig. $4(\mathrm{a})$ ) except near zero rotations about both axes. A further search in $1^{\circ}$ intervals indicated that the minimum $R$ value corresponds, in fact, to $0^{\circ}, 7^{\circ}$, and $0^{\circ}$ rotations about the $x, y$ and $z$-axes, respectively.

A similar search involving translations along $x, y$ and $z$ independently for distances of up to $6 \AA$ gave the lowest $R$ factor for translations of $0 \cdot 0 \pm 0 \cdot 2 \AA$ in all directions. Shifts as small as $0.5 \AA$ yield an $R$ factor $1 \%$ higher than for the zero shifts.

These results are supported by the heavy-atom co-ordinates which we found for the tetrachloroplatinate derivative of demetallized concanavalin A. The two sites, which were determined from Patterson projection maps, have co-ordinates ( $0.00,0 \cdot 06,0.06$ ) and $(0.00,-0.06,0.06)$ to an accuracy of 0.01 of the axial lengths. These co-ordinates are indistinguishable from the heavy-atom positions of this derivative of native concanavalin A (Weinzierl \& Kalb, 1971 ; Hardman et al., 1971; Quiocho et al., 1971) and indicate that any translation must be less than the maximum error in the coordinates (about $1.5 \AA$ ). These heavy-atom co-ordinates are, however, not sensitive to small rotations, and the $7^{\circ}$ rotation about the $y$-axis (located at $x=0, z=0$ ) causes these positions to shift by only $0 \cdot 4 \AA$.

## (b) Refinement

Tables 1 and 2 and Figures 5 and 6 summarize the refinement steps. As we let more parameters vary, such that the ratio of observations to parameters decreased from 2.4 to 1.9 , the refinement remained reasonably well over-determined because of the


Fig. 4. Rotation search for a rigid dimer of native concanavalin A in the crystal lattice of demetallized concanavalin $A$. The ordinate is the agreement factor ( $\left.R=\Sigma| | F_{0}\left|-\left|F_{\mathrm{c}}\right|\right| / \Sigma \mid F_{0}\right) \mid$, where $F_{0}$ and $F_{c}$ represent the observed and calculated structure factors. The abscissa is the rotation angle about the $y$-axis. At rotations of $0^{\circ}$ and $\pm 90^{\circ}$ the space group is actually 1222 rather than $P 2_{1} 22_{1}$, which causes half the reflections to vanish. Therefore, the sum of discrepancies between $F_{o}$ and $F_{c}$ is highest at these angles. The curve is symmetrical with respect to the sign of the rotation angle, since for any particular orientation there will be 2 dimers (constituting a tetramer) rotated clockwise and, because of the space group, another 2 equivalent dimers rotated counterclockwise. (a) For a sample of 83 strong reflections. (b) A close-up in the vicinity of $7 \cdot 0^{\circ}$ rotation, computed from data for 5300 reflections with a minimum spacing of $3 \cdot 2 \AA$.
approximately 6000 distance restraints. It should also be noted that when non-bonded restraints were introduced in step 10 , the $R$ factor did not increase. The average change in phases was $45^{\circ}$ (r.m.s. $60^{\circ}$ ), which is about the same as has been found in other protein refinements (Watenpaugh et al., 1973). A typical cycle, for one of the two monomers, using 7800 reflections, varying 2000 parameters simultaneously, with 1800 atoms and 6000 distance restraints, took 100 minutes and 1000 Kbytes on an IBM 370/168 computer. In retrospect, we realize that the refinement could have proceeded through fewer steps, but in part we were learning how to use the method. The $R$ factor decreased from 0.47 for the unrefined model to 0.26 at the end of refinement. These values are for all 7800 reflections between $10 \AA$ and $3 \cdot 2 \AA$.

## (c) Electron density maps

Part of the difference map calculated for the final structure is shown in Figure 7. The paucity of features is typical of the whole map and is in sharp contrast to Figure 3, which shows a difference map for part of the same region of the unit cell with several incorrectly placed residues. The electron density and refined structure for the entire metal-binding regions of protomers I and II are shown in Figure 8.

The interpretability of both $\left(2\left|F_{0}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right)$ maps and "chopped" difference maps improved dramatically during the refinement. This can be seen by comparing
'TABLE 1
Summary of refinement steps

| Cycle | Refinement operation | $R(\%)$ | No. of reflections | No. of restraints | No. of variables | Comments |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 0 | Generate starting co-ordinates by molecular replacement from native concanavalin A | $42 \cdot 8$ | 5302 | - | - | Data $10 \AA$ to $3 \cdot 2 \AA$ medium and strong reflections |
| 1 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $33 \cdot 7$ | 5302 | 3394 | 3318 | side-chain dihedral angles kept constant |
| 2 | Rigid-body motion of individual amino acids, linked by inter-residuo restraints | $30 \cdot 2$ | 5302 | 3394 | 3318 | Side-chain dihedral angles kept constant |
| 3 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $28 \cdot 7$ | 5302 | 3394 | 3318 | Side-chain dihedral angles kept constant |
| 4 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $33 \cdot 5$ | 7801 | 3394 | 3318 | All data 10 A to $3 \cdot 2 \mathrm{~A}$ side-chain dihedral angles kept constant |
| 5 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $32 \cdot 4$ | 7801 | 3394 | 3318 | All data 10 A to $3 \cdot 2$ A side-chain dihedral angles' kept constant |
| 6 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $31 \cdot 8$ | 7801 | 3394 | 3318 | All data $10 \AA$ to $3 \cdot 2$ A side-chain dihedral angles kept constant |
| 7 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $30 \cdot 4$ | 7801 | 3394 | 3950 | Side-chain dihedral angles $X_{1}$ and $X_{2}$ allowed to vary, except for prolines |
| 8 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $29 \cdot 3$ | 7801 | 3394 | 3950 | Side-chain dihedral angles $X_{1}$ and $X_{2}$ allowed to vary, except for prolines |
| 9 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | 29.0 | 7801 | 3394 | 4050 | Side-chain dihedral angles $X_{1}$ to $X_{4}$ allowed to vary, except for prolines |


| 10 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $28 \cdot 9$ | 7801 | 6158 | 4050 | Side-chain dihedral angles $X_{1}$ to $X_{4}$ allowed to vary, except for prolines. <br> Additional restraints added to minimize repulsive interactions from bad contacts between nonbonded atoms |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 11 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $28 \cdot 2$ | 7801 | 6028 | 4050 | The number of restraints was slightly reduced after refinement caused some non-bonded atoms to move further apart than a fixed minimum distance |
| 12 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | 27.9 | 7801 | 5993 | 4050 | The number of restraints was slightly reduced after refinement caused some non-bonded atoms to move further apart than a fixed minimum distance |
| 13 | Rigid-body motion of individual amino acids, linked by inter-residue restraints | $27 \cdot 6$ | 7801 | 5993 | 4050 | Final cyclo of refinement. Structure factors from this cycle are used for computation of final electron density maps |
| 14 | Group temperature-factor refinement: two per residue, one for the backbone atoms ( $\mathrm{N}, \mathrm{C}_{\alpha}$, $\mathrm{C}_{\beta}, \mathrm{C}^{\prime}$ and O ), and the other for side-chain atoms. (For Gly, Ala and Pro there is only one temperature factor.) All the rest of the parameters are kept constant | $25 \cdot 6$ | 7801 | 0 | 856 | This cycle was computed only for the purpose of establishing the relative flexibility of various groups in the protein and was not used in calculating electron density maps |

Table 2
Some overall properties of the refined structure

| $R$ initial $=0.470$ <br> $R$ final $=0.276$ <br> $R$ group temp. factors $=0.256$ <br> Estimate of error in final co-ordinates $=0.6 \AA \dagger$ |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Mean | r.m.s. | Maximum |  |
| Deviations of corresponding co-ordinates from the initial model (native concanavalin A) $\ddagger$ | Whole structure Backbone Side-chains | $\begin{aligned} & 1.0 \AA \\ & 0.9 \AA \\ & 1.2 \AA \end{aligned}$ | $\begin{aligned} & 1.1 \AA \\ & 1.0 \AA \\ & 1.4 \AA \end{aligned}$ | $\begin{aligned} & 5.0 \AA \\ & 2.4 \AA \\ & 5.0 \AA \end{aligned}$ | (terminal N of $\operatorname{Arg} 172$, protomer I) ( $\mathrm{C}_{a}$ of Gly 98 , protomer I) |
| Deviations of corresponding co-ordinate between protomer I and protomer II§ |  | $1.0 \AA$ | $1 \cdot 2 \mathrm{~A}$ | 4.5 A | ( $\mathrm{C}_{\nu}$ of Val75) |
| Deviations of phase angles from those of the initial model |  | $45^{\circ}$ | $65^{\circ}$ | $180^{\circ}$ |  |
| Deviations from planarity of the peptide bond |  | $4^{\circ}$ | $6{ }^{\circ}$ | $31^{\circ}$ | (Phe213-Ile214, protomer I) |
| Deviations of bond angles in the peptide plane from standard values | tagle ( $\mathrm{C}_{a}-\mathrm{C}-\mathrm{N}$ ) Angle ( $\mathrm{C} \cdot \mathrm{N} \cdot \mathrm{C}_{\alpha}$ ) Angle ( $\mathrm{O}-\mathrm{C} \cdot \mathrm{N}$ ) | $\begin{aligned} & 3^{\circ} \\ & 6^{\circ} \\ & 3^{\circ} \end{aligned}$ | $\begin{aligned} & 4^{\circ} \\ & 8^{\circ} \\ & 3^{\circ} \end{aligned}$ | $\begin{aligned} & 11^{\circ} \\ & 27^{\circ} \\ & 11^{\circ} \end{aligned}$ | (Asn131-Gln132, protomer I) (Ser160-Ser161, protomer I) (Val79-Asp80, protomer II) |
| Deviations of the peptide bond length from a standard value |  | $<0.01$ A | 0.01 A | 0.04 A |  |

[^0]

Fig. 5. Decrease of $R=\Sigma| | F_{\mathrm{o}}\left|-\left|F_{\mathrm{c}}\right|\right| / \Sigma\left|F_{\mathrm{o}}\right|$ during the reciprocal-space refinement of demetallized concanavalin A. In the first 3 cycles a partial data set of 5300 reflections was used. From cyele no. 4 the entire data set of 7800 reflections was used. The initial $R$ factor for the full data set was 0.47 as indicated by the dashed line. For details of each refinement cycle see Table 1.


Fig. 6. Mean phase change of the calculated structure factors from those of the starting model, during the reciprocal-space refinement of demetallized concanavalin $A$. The dashed line indicates the incorporation of the entire data set.

Figure 2 with Figure 8(a) (middle and bottom) which show the same portion of map before and after refinement. In general, both types of map are very similar, but occasionally the chopped difference map lacks density where there is density in the other map and the difference map is relatively clean. Therefore, $\left(2\left|F_{o}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{c}\right)$ maps were usually used for interpretation.

In general, the polypeptide chain can be traced reasonably well, but occasionally there is lack of density, for example at Pro20 in protomer I, and between residues 14 and 17 and at Arg33 in protomer II. Since in the difference map these features appear in troughs with no unmatched peaks nearby, we believe that they represent flexible segments of the structure. On the other hand, there are some unaccounted for extra peaks, both in the $\left(2\left|F_{\mathrm{o}}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right)$ mapand inthe $\left(\left|F_{\mathrm{o}}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right)$ difference map. Most of these are clearly outside the protein region, but a


Fig. 7. A 5 A-thick portion of the difference electron density map for the refined structure of demetallized concanavalin $A$ in the metal-binding region of protomer $I$, sectioned along the z-axis. The coefficients of the Fourier summation are $\left(\left|F_{o}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right)$. Positive and negative contours are shown in solid and dashed lines, respectively, at equal intervals of twice the standard deviation ( $\sigma$ ) of the non-protein region of the map. Superimposed is the atomic model. Residuc numbers designated with a superscript are from a neighbouring moleculo.
possible exception is the peak situated near Ser2l (see Figs 7, 8(a) bottom, and 8(b) bottom). At an intermediate stage of the interpretation process, we tried to account for this density by an extended loop of residues 18 to 22 . The difference map shown in Figure 3 clearly indicates this preliminary interpretation to be incorrect. We therefore believe this and the other extra peaks may represent ions or ordered solvent.

The group temperature-factor refinement provides a cross check on weak density features. The temperature factors, in fact, proved to be very sensitive to any lack of correspondence between atomic positions and electron density. Sometimes high temperature factors indicated only minor imperfections such as the ends of the sidechains of Ile4 (Fig. 8(b), top) or Ile29 (Fig. 8(a), middle) lying just on the edge of electron density. In other cases however they indicated what are evidently flexible segments in the structure which fall in troughs in the difference maps. As might be expected, most of the unusually high temperature factors were found to correspond to structural features on the surface of the molecule. Particularly high temperature factors were found for residues 118 to 122 in an external loop which is often found to be cleaved in the native structure (Edelman et al., 1972) and for the carboxy-terminal residues 235 to 237 .

## (d) The structure of the metal-binding region, residucs 7 to 35

The folding of the backbone in both protomers is very similar to that in native concanavalin A, but there is a slight, yet consistent, closing up around the "metal positions" (see Fig. 9). The more striking conformational differences are confined to Aspl9 and His24, which are located closer to each other than in the native form. The
distance between the carbon atom of the Asp19 carboxyl group and either of the two nitrogen atoms of the imidazole ring of His 24 diminishes by about $1 \AA$ in both protomers. This is due to changes in the positions of both Asp19 and His24. However, the position of Asp19 is different in the two protomers. In protomer I, Aspl9 is pointing away from the calcium position, thus increasing the $\mathrm{Ca}-\mathrm{C}_{\gamma}$ (Aspl9) distance from $3.5 \AA$ to $5.5 \AA$, whereas in protomer II this distance is roughly the same as in native concanavalin A.
Some parts of the structure have unusually high temperature factors accompanied by weak density in the map and negative regions in the $\left(\left|F_{\mathrm{o}}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right) \mathrm{map}$, with no unaccounted peak nearby. We have interpreted such features as disorder. Examples are the side-chain of Glu8 beyond $\mathrm{C}_{\gamma}$, which in the native form is liganded to the transition metal, and the external side-chains in the loop of residues 11 to 23

## 4. Discussion

This study has been carried out at what is considered to be an intermediate resolution $(3 \cdot 2 \AA)$ and so, before proceeding to a discussion of the results, it is necessary to consider the reliability of an analysis made at this resolution. In fact, it has been shown from a number of structure determinations at about this resolution that it is quite adequate for visualising the polypeptide backbone and most side-chains (Matthews, 1977). Molecular replacements at intermediate resolutions have been used successfully to elacidate conformational differences between similar structures in different crystal forms (Anderson, 1975; Fehlhammer et al., 1975; Moult et al., 1976; Schmid \& Herriott, 1976), and the CORELS refinement procedure, when applied to a tRNA at $2 \cdot 7 \AA$ resolution, proved very effective in clarifying a number of structural details (Sussman et al., 1978).

It is clear that, in the present work too, this refinement procedure has greatly improved the clarity and interpretability of the maps (cf. Figs 2 and 8) to the point where individual side-chains can generally be distinguished. Inspection of the maps provides the best check of the reliability of the interpretation. Comparison of a difference map including an incorrectly positioned length of chain (Fig. 3) with that for the final structure (Fig. 7) indicates that major errors in the model would be immediately apparent. The $\left(2\left|F_{0}\right|-\left|F_{\mathrm{c}}\right|\right) \exp \left(i \alpha_{\mathrm{c}}\right)$ maps (Fig. 8) show that nearly all the atoms are positioned centrally within strong electron density which is well localized so that quite small changes in co-ordinates ( $\sim 0.5 \AA$ ) would worsen this fit. For atoms so situated there can be little doubt of the correctness of the model. For the few regions where electron density and atomic positions are not so well matched the situation is less clear. Portions of the structure in weak density also have high temperature factors and occur on the surface of the molecule or other places where flexibility might reasonably be expected (cf. Results, section (c)). The very few peaks not assigned to atoms are generally much weaker than the rest of the density and outside the protein region (see Fig. 8) as might be expected for solvent or ions. Though we cannot be certain of their interpretation in every case the paucity and sparse distribution of the poorly matched regions in the maps would effectively rule out major misinterpretations in our model, though the possibility of some local errors cannot be excluded.

The presence of two crystallographically independent monomers, which turn out to have closely similar structures, also provides a useful check of our model. We have


Fig. 8. Three successive $5 \AA$-thick portions of the eloctron density map for the refined structure of demetallized concanavalin $A$ in the metal-binding region, sectioned along the z-axis. The coefficients of the Fourier summation are ( $\left.2\left|F_{0}\right|--\mid F_{0}\right) \exp \left(i \alpha_{0}\right)$. Contours are drawn from a level corresponding to the standard deviation (o) in the mon-protein part of the map in oqual intervals of half that value. Superimposed is the atomic model. Residue numbers designated with

(b)

a superseript are from a neighbouring molecule. (a) Protomer I. (b) Protomer II. Note the high correlation between lack of density here and the corresponding negative density in the difference map shown in Fig. 7. See, for example, Glu8 and Pro20 in Protomer I. We have intorpreted such features as representing disorder.


Fig. 9. Stereo representation of residues 7 to 35 in demetallized and native concanavalin i viewed down the $z$-axis. The backbone is represented by virtual bonds between successive alphacarbons. Details are shown of ligands which participate in metal binding, including the carbonyl group of Tyrl2 and side-chains of Glu8, Asp10, Asn14, Asp19, His24 and Ser34. The native structure is the initial starting model used in the refinement process (see text). Flexible parts are shown in dashed lines. (a) Protomer I (solid lines) and native (open lines). (b) Protomer II (solid lines) and native (open lines), rotated by $180^{\circ}$ to have roughly the same orientation as protomer I.
tried to estimate the likely error in the co-ordinates by comparing these for equivalent atoms in the monomers excluding those groups which are particularly flexible or might have different conformations because of different intermolecular contacts. The r.m.s. difference in these co-ordinates is $0.6 \AA$, whereas the r.m.s. co-ordinate shift effected by the refinement is about twice this value (Table 2).

In establishing the differences between the native and demetallized structures we have been hindered by the current unrefined state of the structure of the former. However, several significant comparisons do emerge. The most striking feature of the demetallized structure is the close similarity to the native protein. Main-chain differences are small ( $2 \cdot 4 \AA$ at most), there are only a few side-chain rearrangements. and the level of differences for the metal-binding region is similar to that of the rest of the structure. There is some local disorder in the backbone which may explain the greater susceptibility of the demetallized protein to proteolysis (Blumberg \& Tal, 1976). Larger differences might have been expected in view of the loss of sugar-binding capacity upon removal of the metals (Yariv et al., 1968), even though the sugarbinding site seems to be about $10 \AA$ from the $\mathrm{Mn}^{2+}$-site (Brewer et al., 1973; Villafranca \& Viola, 1974; Alter \& Magnuson, 1974). In the following discussion of
differences between the structures we concentrate on one of the two protomers (I) and comment on the differences between protomers at the end.

The $\mathbf{M n}^{2+}$ ligands on the protein (Glu8, Asp10, Asp19 and His24) show the most difference. The side-chain of Glu8 becomes disordered and the distance from the carboxyl group of Aspl9 to the imidazole ring of His24 shortens substantially (see Fig. 9). These changes are most easily explained in terms of the changes in charge in the region on removing the metals. For the demetallized structure Glu8, Asp10 and Asp19 may be expected to be negatively charged, and His24 to be positively charged at the pH of crystallization $(5 \cdot 0)$. Thus, in the demetallized structure, charge-charge interactions stabilize the relationship between Asp10, Asp19 and His24, while Glu8, in the absence of an interaction with the metal or any other stabilizing interaction, becomes disordered. The involvement of Aspl9 in a strong interaction is supported by the higher density and lower temperature factor of the side-chain of this residue compared with its backbone (see Fig. 8(a), middle, and 8(b), middle). The presence of a positively charged His24 is also borne out by the report of Gachelin et al., (1972) that two protons are lost on metal-binding. However, we cannot unequivocally account for the second proton.

This rearrangement of residues 10,19 and 24 suggests an explanation for the metalfree protein's inability to bind calcium. Aspl 9 is $2 \AA$ further away from the Ca-site than in the native structure (see Figs 9(a) and 10), substantially worsening the liganding situation. However, when a transition metal ion is bound in site S1, His24 loses its positive charge, and the carboxyl groups of Aspl0 and Aspl9 would presumably be reoriented in accordance with the new charge distribution, creating site $\$ 2$ for $\mathrm{Ca}^{2+}$ binding. This view is supported by the report that calcium can be bound even to the Mn-free protein at pH 7 , when the histidine would be uncharged (Alter et al., 1977).

In protomer II these rearrangements are not so pronounced (see Fig. 9(b)). However, there are indications of possible additional backbone disorder (see residues 14 to 17 in Fig. 8(b), top), and there is also a close contact with a neighbouring molecule in the region of residue 16 not present in the other protomer. We therefore believe that the protomer I structure better represents the situation in solution.

It should be noted that our results disagree with those of Becker et al. (1976), who report an opening of the loop of residues 11 to 23 in the demetallized structure. We can find no evidence for such a change in our maps after refinement. In a recent paper (Reeke et al., 1978) this group has also reported that the peptide linkage between Ala207 and Asp208, which was assigned a cis configuration in native concanavalin A, is trans in the demetallized structure, and they have suggested that this might be related to the measurement, from a kinetic analysis of nuclear magnetic resonance data, of a $22 \mathrm{kcal} \mathrm{mol}^{-1}$ energy barrier in the activation of the demetallized protein (Brown et al., 1977). However, careful examination of these regions of the structure in both protomers using difference and chopped difference maps have shown us that cis configurations fit our electron density maps very well (Fig. 10). The trans: configurations we have tried to fit to the maps have proved much less satisfactory.

The structural reason for the inability of demetallized concunavalin A to bind saccharides is currently difficult to explain, as the sugar ligands have not been unambiguously identified. However, residues likely to be involved in saccharide binding have been suggested from the location of iodine atoms in iodinated sugar derivatives associated with concanavalin A (Becker et al., 1976; Hardman \& Ainsworth, 1976).


Fig. 10. Stereoscopic view of a $12 \AA$-thick portion of a "chopped" difference electron density map for the refined structure of demetallized concanavalin $A$ in the region of protomer I near residues 20 through 26 as well as His205, Hro206, Ala207, Asp208, Gly $209,11 \mathrm{e} 210$ and Ala211. The coefficients of the Fourier summation are $\left(\left|F_{o}\right|-\left|F_{c}\right|\right) \exp \left(i \alpha_{c}\right)$ with all residues but 205 through 211 of protomers I and II included in the calculation of structure-factor amplitudes and phases. Contours are drawn at a level corresponding to 2 standard deviations ( $\sigma$ ) in the whole map. Superimposed is the refined structural model. Note that there is no appreciable electron density except in the portion of the map corresponding to the 7 residues omitted from the structurefactor calculations and that this fits well with the superimposed structure having a cis configuration for the peptide linkage between Ala207 and Asp208 and a trans configuration for all the others.

Among these possible saccharide ligands, the positions of the hydroxyls of Tyrl2 and Tyrl00 deviate by roughly $1.5 \AA$ from their locations in native concanavalin $A$, and the positions of carboxyl carbon atoms of Asp 16 and Asp208 deviate by $1.0 \AA$ and of carbon $\mathrm{C}_{\mathrm{z}}$ of Arg228 by $1.5 \AA$. These differences could perhaps account for the lack of binding capacity for saccharides. A detailed evaluation of the conformational changes in sugar ligands will however have to await the elucidation of the structure of a concanavalin A-saccharide complex.

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## REFERENCES

Alter, G. M. \& Magnuson, J. A. (1974). Biochemistry, 13, 4038-4045.
Alter, G. M., Pandolfino, E. R., Christie, I. J. \& Magmuson, J. A. (1977). Biochemistry. 16, 4034-4038.
Anderson, L. (1975). J. Mol. Biol. 94, 33-49.
Barber, B. H. \& Carver, J. P. (1973). J. Biol. Chem. 248, 3353~3355.
Becker, J. W., Reeke, G. N. Jr, Wang, J. L., Cunningham, B. A. \& Edelman, G. M. (1975). J. Biol. Chem. 250, 1513-1524.

Becker, J. W., Reeke, G. N. Jr. Cunningham, B. A. \& Edelman, G. M. (1976). Nature (London), 259, 406-409.
Blumberg, S. \& Tal, N. (1976). Biochim. Biophys. Acta, 453, 357-364.
Brewer, C. F., Sternlicht, H., Marcus, D. M. \& Grollman, A. P. (1973). Proc. Nat. Acad. Sci., U.S.A. 70, 1007-1011.
Brown, R. D., Brewer, C. F. \& Koenig, S. H. (1977). Biochemistry, 16, 3883-3896.
Diamond, R. (1974). J. Mol. Biol. 82, 371-391.
Edelman, G. M., Cunningham, B. A., Reeke, G. N. Tr, Becker, J. W., Waxdal, M. J. \& Wang, J. L. (1972). Proc. Nat. Acad. Sci., U.S.A. 69, 2580-2584.
Fehlhammer, H., Schiffer, M., Epp, O., Colman, P. M., Lattman, E. E., Schwager, P., Steigemann, W. \& Schramm, H. J. (1975). Biophys. Struct. Mechanism, 1, 139-146.
Gachelin, G., Goldstein, L., Hofnung, D. \& Kalb, A. J. (1972). Eur.J. Biochem. 30, 155-162.
Goldammer, E. von \& Zorn, H. (1974). Eur. J. Biochem. 44, 195-199.
Hardman, K. D. \& Ainsworth, C. F. (1972). Biochemistry, 11, 4910-4919.
Hardman, K. D. \& Ainsworth, C. F. (1975). The Irotein Data Bank. Brookhaven National Laboratory, Upton, New York.
Hardman, K. D. \& Ainsworth, C. F. (1976). Biochemistry, 15, 1120-1128.
Hardman, K. D., Wood, M. K., Schiffer, M., Edmundson, A. B. \& Ainsworth, C. F. (1971). Proc. Nat. Acad. Sci., U.S.A. 68, 1393-1397.
Jack, A., Weinzierl, J. \& Kalb, A. J. (1971). J. Mol. Biol. 58, 389-395.
Kalb, A. J. \& Levitzki, A. (1968). Biochem. J. 109, 669-672.
Kalb, A. J. \& Pecht, I. (1973). Biochim. Biophys. Acta, 303, 264-268.
Isuzzati, V. (1953). Acta Crystallogr. 6, 142-152.
Matthows, B. W. (1977). In The Proteins (3rd edition) (Neurath H. \& Hill, R. L., ods), vol. 3, pp. $445-446$, Academic Press, New York.
Meirovitch, E. \& Kalb, A. J. (1973). Biochim. Biophys. Acta, 303, 258-263.
Meirnvitch, F., L.117, 7. \& Kalb (Gilboa), A. J. (1974). J. Amer. Chem. Soc. 96, 7542-7546.
Moult, J., Yonath, A., Traub, W., Smilansky, A., Podjarny, A., Rabinovich, D. \& Saya, A. (1976). J. Mol. Biol. 100, 179-195.

Nicolau, (., Kalb, A. J. \& Yariv, J. (1969). Biochim. Biophys. Acta, 194, 71-73.
North, A. C. T., Phillips, D. C. \& Mathews, F. S. (1968). Acta Crystallogr. sect. A. 24, 351.359.

Nyburg, S. C. (1974). Acta Crystallogr. sect. B, 30, 251-253.
Quiocho, F. A., Reeke, G. N. Jr, Becker, J. W., Lipscomb, W. N. \& Edelman, G. M. (1971). Proc. Nat. Acad. Sci., U.S.A. 68, 1853-1857.
Reeke, G. N. Jr, Becker, J. W. \& Edelman, G. M. (1975). .J. Biol. Chem. 250, 1525-1547.
Reeke, G. N. Jr, Becker, J. W. \& Edelman, G. M. (1978). Proc. Nat. Acad. Sci., U.S.A. 75, 2286-2290.
Rollett, J. S. \& Sparks, R. A. (1960). Acta Crystallogr. 13, 273-274.
Rossmann, M. G. (1972). In The Molecular Replacement Method, International Science Review Series, vol. 13, Gordon and Breach, New York.
Schmid, M. F. \& Herriott, J. R. (1976). J. Mol. Biol. 103, 175-190.
Shoham, M., Kalb, A. J. \& Pecht, L. (1973). Biochemistry, 12, 1914-1917.
Shoham, M., Sussman, J. L., Yonath, A., Moult, J., Traub, W. \& Kalb (Gilboa), A. J. (1978). FEBS Letters, 95, 54-56.

Sumner, J. B. (1919). J. Biol. Chem. 37, 137-142.
Sumner, J. B. \& Howell, S. F. (1936). .J. Biol. Chem. 115, 583-588.
Sussman. J. L., Holbrook, S. R., Church, G. M. \& Kim. S. H. (1977). Acta Crystallogr. sect. $A, 33,800-804$.
Sussman, J. L., Holbrook, S. R.. Warrant, R. W.. (hurch, G. M. \& Kim, S. H. (1978). J. Mol. Biol. 123, 607-630.

Villafranca, J. J. \& Viola, R. E. (1974). Arch. Biochem. Biophys. 160, 465-468.
Watenpaugh, K. D., Sieker, L. C., Herriott, J. R. \& Jensen, L. H. (1973). Acta Crystallogr. sect. $B, 29,943-956$.
Watson, H. C., Shotton. D. M., Cox, J. M. \& Muirhead, H. (1970). Nature (London), 225, 806-811.
Weinzierl, J. \& Kalb, A. J. (1971). FEBS Letters, 18, 268-270.
Yariv, J.. Kalb, A. J. \& Levitzki, A. (1968). Biochim. Biophys. Acta. 165, 303-305.


[^0]:    (res 235-237) were excluded for the purpose of this calculation.
    $\ddagger$ Co-ordinates of native concanavalin A from Hardman \& Ainsworth (1975).
    $\S$ With the aid of a computer program to superimpose co-ordinates of protomer II onto those of protomer 1 in anch a way as to minimize their enrrosponding deviations (Nyburg, 1974). All atoms included in calculation.

    T The only bond length altered in the refinement process is the peptide linkage botween successive residues, and the only bond angles varied are within the peptide plane. Standard bond lengths and bond angles according to Diamond (1974).

