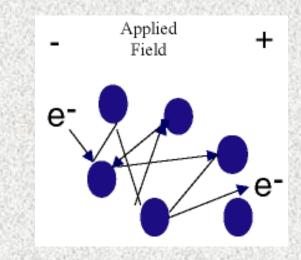
# **Electronic properties**





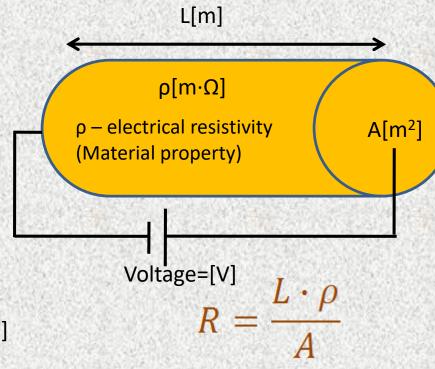
Primer Materials For Science Teaching

#### Ohm's low

$$I = rac{V}{R}$$
 I – current [A] V – Voltage [V] R – Resistance [ $\Omega$ ]

Power dissipation:
$$P = I^{2}R = V \cdot I = \frac{V^{2}}{R}$$
P-Watt [W]

$$\sigma = 1/\rho$$
  $\sigma$  – electrical conductivity [S/m]



Resistance has geometrical dependence

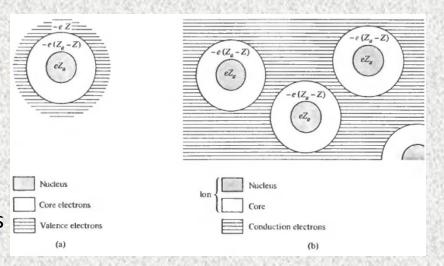
$$J = I/A$$
 J – Current density (flux)

$$E=V/L$$
 E – electric field [V/m]

$$J = \sigma \cdot E$$

#### Drude model

n – electron density
n<sub>i</sub> – metal ion density
e – electron charge
m<sub>e</sub>- electron mass
μ- mobility
a- atomic (metallic) radius



#### **Assumptions**

- •All atoms gives their valance electron to the 'sea of electron'
- •Ion are localized, Electron moves in straight lines between collisions
- •Relaxation time: τ time between collisions

$$\sigma = \mu n e$$
  $\sigma = \frac{n e^2 \tau}{m_e}$  The electrical conductivity related to the relaxation time  $\mu = \frac{e \tau}{m_e}$   $\tau = \frac{m_e}{\rho n e^2} = \frac{1}{n_i \pi a^2} \cdot \sqrt{\frac{m_e}{3KT}}$ 

# Typical relaxation times

Table 1.2 ELECTRICAL RESISTIVITIES OF SELECTED ELEMENTS<sup>a</sup>

$(\rho/T)_{373}$						
ELEMENT	77 K	273 K	373 K	$\frac{(\rho/T)_{273 \text{ K}}}{(\rho/T)_{273 \text{ K}}}$		
Li	1.04	8.55	12.4	1.06		
Na	0.8	4.2	Melted			
K	1.38	6.1	Melted			
Rb	2.2	11.0	Melted			
Cs	4.5	18.8	Melted			
Cu	0.2	1.56	2.24	1.05		
Ag	0.3	1.51	2.13	1.03		
Au	0.5	2.04	2.84	1.02		
Be		2.8	5.3	1.39		
Mg	0.62	3.9	5.6	1.05		
Ca		3.43	5.0	1.07		
Sr	7	23				
Ba	17	60				
Nb	3.0	15.2	19.2	0.92		
Fe	0.66	8.9	14.7	1.21		
Zn	1.1	5.5	7.8	1.04		
Cd	1.6	6.8				
Hg	5.8	Melted	Melted			
Al	0.3	2.45	3.55	1.06		
Ga	2.75	13.6	Melted			
In	1.8	8.0	12.1	1.11		
TI	3.7	15	22.8	1.11		
Sn	2.1	10.6	15.8	1.09		
Pb	4.7	19.0	27.0	1.04		
Bi	35	107	156	1.07		
Sb	8	39	59	1.11		

Resistivities in microhm centimeters are given at 77 K (the boiling point of liquid nitrogen at atmospheric pressure), 273 K, and 373 K. The last column gives the

[μΩ·cm] = 1X10<sup>-8</sup>[Ω ·m]

Longmans Green, London, 1966.

approximate linear temperature ture.

Physical and Chemical Constants,

Table 1.3				
DRUDE RELAXATION TO	MES IN	UNITS	OF 10	14 SECOND

ELEMENT	77 K	273 K	373 K
Li	7.3	0.88	0.61
Na	17	3.2	
K	18	4.1	
Rb	14	2.8	
Cs	8.6	2.1	
Cu	21	2.7	1.9
Ag	20	4.0	2.8
Au	12	3.0	2.1
Be		0.51	0.27
Mg	6.7	1.1	0.74
Ca		2.2	1.5
Sr	1.4	0.44	
Ba	0.66	0.19	
Nb	2.1	0.42	0.33
Fe	3.2	0.24	0.14
Zn	2.4	0.49	0.34
Cd	2.4	0.56	
Hg	0.71		
AI	6.5	0.80	0.55
Ga	0.84	0.17	7.15.To.
In	1.7	0.38	0.25
TI	0.91	0.22	0.15
Sn	1.1	0.23	0.15
Pb	0.57	0.14	0.099
Bi	0.072	0.023	0.016
Sb	0.27	0.055	0.036

<sup>&</sup>quot;Relaxation times are calculated from the data in Tables 1.1 and 1.2, and Eq. (1.8). The slight temperature dependence of n is ignored.

# **Drift velocity**

V<sub>th</sub> – thermal velocity E – electric field V<sub>drift</sub> – drift velocity

$$V_{th} = \sqrt{\frac{3KT}{m_e}}$$

$$J = \sigma E = nev_{drift}$$

- ➤ When J = 0 the drift velocity is zero. Meaning that in average each electron has zero displacement
- When J ≠ electron moves with typical velocity of v<sub>drift</sub>

#### Electron moves in metal

# Example 1

You apply a potential difference of 4.5 [V] between the ends of a wire that is 2.5 [m] in length and 0.64 [mm] in radius. The resulting current through the wire is 18 [A]. What is the resistivity of the wire? What is the heat loss?

$$R = V/I = 0.25[\Omega]$$

$$A = \pi r^2 = 1.29X10^{-6}$$

$$\rho = \frac{R \cdot A}{L} = 12.9X10^{-8} [\Omega \cdot m]$$

$$P = I \cdot V = I^2 \cdot R = 81 [W]$$

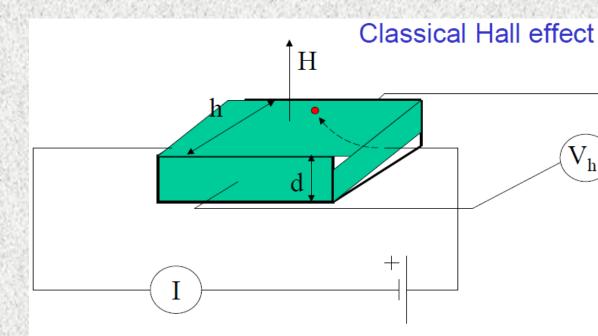
# Example 2

The resistivity of Cu is  $1.7 \times 10^{-8} \Omega m$  at 300 K and the electron density is  $8.5 \times 10^{28} \text{ m}^{-3}$ .

- (a) Calculate the relaxation time of electrons in Cu at 300 K.
- (b) Calculate the mean free path of the electrons using Drude approximation.

a) 
$$\tau := \frac{me}{\rho \cdot n \cdot e} 2 \qquad \tau = 2.46 \times 10^{-14} s \qquad \begin{array}{l} m_e = 9.1093837 \times 10^{-31} \ \text{Kg} \\ e = 1.60217663 \times 10^{-19} \ \text{C} \end{array}$$

b) Vther := 
$$\left(\frac{3 \cdot k \cdot T}{me}\right)^{\frac{1}{2}}$$
 lther := Vther  $\cdot \tau$   $\frac{lther}{10^{-9} \cdot m} = 2.874$  nm



 $V_h$  Lorentz force:  $F_L = e \text{ v} \times H$ ;

Electrostatic force:  $F_E$ =Ee= $V_he/l$ 

At equilibrium  $F_L = F_E$  and

$$v \times H = V_h/h$$

 $V_h = \frac{HI}{end} = R_h H \frac{I}{d}$ 

Current density is related to the velocity as ven=I/(dh) =>

$$R_h = \frac{1}{en}$$

 $R_h$  is Hall constant

Classical theory predicts that -R<sub>h</sub>en=1

#### **Example 3**

Prove that the combination of Hall effect measurements and resitivity measurements permits determination of the electron relaxation time

$$R_{h} = \frac{1}{e \cdot n}$$
Hall voltage  $V_{h} = R_{h} \cdot \frac{IH}{d} = \frac{H \cdot I}{e \cdot n \cdot d} \Rightarrow n = \frac{H \cdot I}{e \cdot d \cdot V_{h}}$ 

Conductivity 
$$\sigma = \frac{ne^2\tau}{m_e} \Rightarrow \tau = \frac{m_e\sigma}{ne^2} = \frac{m_e \cdot \sigma \cdot d \cdot V_h}{H \cdot I \cdot e}$$